

## A NEW OLEANOLIC GLYCOSIDE FROM *POLYSCIAS SCUTELLARIA*

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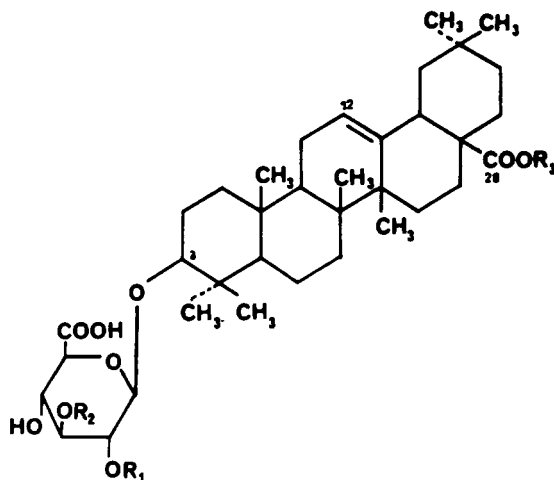
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**ABSTRACT.**—A new triterpenoid saponin, named polysciasaponin P<sub>1</sub> [**8**], has been isolated from *Polyscias scutellaria* leaves and identified as 3-O-[β-D-glucopyranosyl(1→4)-β-D-glucopyranosyl(1→2)-β-D-glucuronopyranosyl] oleanolic acid 28-O-β-D-glucopyranoside. The structure was established by chemical and spectroscopic means (fabms, <sup>13</sup>C nmr, gc-ms).

In previous work, we reported the isolation and structure determination of polysciasaponins P<sub>7</sub> [**2**], P<sub>6</sub> [**3**], P<sub>4</sub> [**5**] (1), P<sub>5</sub> [**4**], and P<sub>2</sub> [**7**] (2) from the leaves of *Polyscias scutellaria* (Burm.f.) Fosb (Araliaceae) used in the islands of the Western Pacific Ocean, especially in Vanuatu, as an anti-inflammatory. In the present paper we report the isolation and structure elucidation of polysciasaponins P<sub>3</sub> [**6**] and P<sub>1</sub> [**8**]; P<sub>1</sub> is a new oleanolic glycoside.

An EtOH extract of dried leaves was fractionated. The *n*-BuOH fraction yielded a crude mixture of saponins. Separation of the *n*-BuOH extract by cc and preparative tlc on Si gel afforded pure saponins P<sub>1</sub> and P<sub>3</sub>.

On acid hydrolysis, polysciasaponins P<sub>1</sub> and P<sub>3</sub> afforded oleanolic acid [**1**] as the aglycone, identified by comparison with a standard sample (tlc, ms, and gc-ms). The sugars obtained from the hydrolysates were identified by tlc and gc



- 1 OH at C-3, R<sub>3</sub>=H
- 2 R<sub>1</sub>=R<sub>2</sub>=R<sub>3</sub>=H
- 3 R<sub>1</sub>=R<sub>3</sub>=H, R<sub>2</sub>=Glc
- 4 R<sub>2</sub>=R<sub>3</sub>=H, R<sub>1</sub>=Glc
- 5 R<sub>1</sub>=H, R<sub>2</sub>=R<sub>3</sub>=Glc
- 6 R<sub>2</sub>=H, R<sub>1</sub>=R<sub>3</sub>=Glc
- 7 R<sub>2</sub>=R<sub>3</sub>=H, R<sub>1</sub>=Glc(1→4)Glc
- 8 R<sub>2</sub>=H, R<sub>1</sub>=Glc(1→4)Glc, R<sub>3</sub>=Glc

Glc=β-D-glucopyranosyl

as glucuronic acid and glucose for P<sub>1</sub> (1:3), glucuronic acid and glucose for P<sub>3</sub> (1:2).

The mol wt and the sugar sequence of P<sub>1</sub> and P<sub>3</sub> were established by fabms in negative ion mode. The following signals were observed: *m/z* 1117 for P<sub>1</sub>, *m/z* 955 for P<sub>3</sub> corresponding to deprotonated molecular ion [M - H]<sup>-</sup>; *m/z* 955 for P<sub>1</sub> and *m/z* 793 for P<sub>3</sub> resulting from the loss of a glucosyl moiety [M - H - 162]<sup>-</sup>.

On mild acid hydrolysis, P<sub>1</sub> initially gave P<sub>2</sub>, P<sub>3</sub>, P<sub>5</sub>, then P<sub>7</sub>; polysciasaponin P<sub>3</sub> initially gave P<sub>5</sub> and P<sub>7</sub>.

Basic hydrolysis of polysciasaponin P<sub>1</sub> yielded P<sub>2</sub>, 3-*O*-[β-D-glucopyranosyl(1→

4)-β-D-glucopyranosyl(1→2)-β-D-glucuronopyranosyl] oleanolic acid; basic hydrolysis of P<sub>3</sub> afforded P<sub>5</sub>, 3-*O*-[β-D-glucopyranosyl(1→2)-β-D-glucuronopyranosyl] oleanolic acid. These results indicated that polysciasaponins P<sub>1</sub> and P<sub>3</sub> are ester glycosides of P<sub>2</sub> and P<sub>5</sub>, respectively. The structure of polysciasaponin P<sub>2</sub> was established as 3-*O*-[β-D-glucopyranosyl(1→4)-β-D-glucuronopyranosyl] oleanolic acid (1), and polysciasaponin P<sub>5</sub> as 3-*O*-[β-D-glucopyranosyl(1→2)-β-D-glucuronopyranosyl] oleanolic acid (4). Upon comparison of the <sup>13</sup>C-nmr spectrum of P<sub>1</sub> with that of P<sub>2</sub> (Table 1), all of the carbon signals of the

TABLE 1. <sup>13</sup>C nmr Spectral Data of Polysciasaponins P<sub>1</sub> [8] and P<sub>2</sub> [7] in C<sub>5</sub>D<sub>5</sub>N (δ ppm/TMS).

Carbon	Polysciasaponin P <sub>1</sub>	Polysciasaponin P <sub>2</sub>
<b>Oleanolic acid</b>		
C-3	89.52	89
C-12	122.10	122.2
C-13	144.12	144.52
C-28	176.28	179.9
	(-3.6 ppm)	
<b>Glucuronic acid</b>		
C-1'	104.40	104.6 <sup>g</sup>
C-2'	83.26	82.9
C-3'	76.43 <sup>c</sup>	76.4 <sup>h</sup>
C-4'	71.85 <sup>c</sup>	71.15
C-5'	76.46 <sup>c</sup>	76.1 <sup>h</sup>
C-6'	174.35	174.4
<b>Glucose</b>		
C-1''	104.71 <sup>a</sup>	104.4 <sup>g</sup>
C-2''	73.91 <sup>b</sup>	74.5
C-3''	78.10 <sup>d</sup>	77.20 <sup>i</sup>
C-4''	80.97	80.5
C-5''	77.63 <sup>d</sup>	78.07 <sup>i</sup>
C-6''	62.40 <sup>f</sup>	62.2
<b>Glucose</b>		
C-1'''	104.67 <sup>a</sup>	104.3 <sup>g</sup>
C-2'''	74.71 <sup>b</sup>	74.5
C-3'''	77.63 <sup>d</sup>	77.25 <sup>i</sup>
C-4'''	71.34 <sup>e</sup>	71.16
C-5'''	77.58 <sup>d</sup>	77.85 <sup>i</sup>
C-6'''	62.31 <sup>f</sup>	61.9 <sup>j</sup>
<b>Glucose</b>		
C-1''''	95.54	
C-2''''	73.74 <sup>b</sup>	
C-3''''	78.73 <sup>d</sup>	
C-4''''	71.90 <sup>e</sup>	
C-5''''	78.47 <sup>d</sup>	
C-6''''	62.97 <sup>f</sup>	

<sup>a-j</sup>Signals with the same superscript are interchangeable.

aglycone and glucosyl moieties appeared at almost the same positions, demonstrating that  $P_1$  is an ester glycoside of  $P_2$ . The chemical shift of C-28 of the aglycone moiety of  $P_1$  was displaced downfield by 3.6 (176.3 ppm for  $P_1$ ; 179.9 ppm for  $P_2$ ), indicating that the carboxy group is esterified with a glucosyl moiety.

The second molecule of glucose in polysciasaponin  $P_3$  might be linked either with COOH-28 of oleanolic acid or COOH-6' of glucuronic acid. Methylated derivative of  $P_3$  obtained with  $\text{CH}_2\text{N}_2$  was successively hydrolyzed with  $\beta$ -glucosidase and  $\beta$ -glucuronidase. Oleanolic acid, and not an aglycone methyl ester, was obtained, clearly indicating that one  $\beta$ -D-glucose is linked to the C-28 at the carboxylic acid of oleanolic acid.

In conclusion, the structure of polysciasaponin  $P_1$  was established as 3-O- $[\beta$ -D-glucopyranosyl (1 $\rightarrow$ 4)- $\beta$ -D-glucopyranosyl (1 $\rightarrow$ 2)- $\beta$ -D-glucuronopyranosyl] oleanolic acid-28-O- $\beta$ -D-glucopyranoside [**8**] and the structure of polysciasaponin  $P_3$  as 3-O- $[\beta$ -D-glucopyranosyl (1 $\rightarrow$ 2)- $\beta$ -D-glucuronopyranosyl] oleanolic acid 28-O- $\beta$ -D-glucopyranoside [**6**]. Polysciasaponin  $P_3$  is identical with chikusetsusaponin V, isolated from *Panax* species (3-9).

## EXPERIMENTAL

### GENERAL EXPERIMENTAL PROCEDURES.—

Gc-ms was performed on a SE54 coated quartz capillary column with a Hewlett-Packard in electronic impact mode. Fabms were obtained on a VG Micromass ZAB-HF spectrometer in the negative ion mode; samples were suspended in polyethylene glycol;  $^{13}\text{C}$ -nmr spectra were recorded on a Bruker WP 200 in  $\text{C}_5\text{D}_5\text{N}$ . The chemical shifts are given in ppm. TMS was used as internal standard. Tlc was carried out on Si gel 60 (Merck). The tlc systems employed were: for saponins *n*-BuOH-HOAc-H<sub>2</sub>O (4:1:1) (system 1), EtOAc-MeOH-H<sub>2</sub>O (20:5:2) (system 2), and  $\text{CHCl}_3$ -MeOH-H<sub>2</sub>O (30:15:2) (system 3); and for aglycones  $\text{CHCl}_3$ -MeOH (97:3) (system 4) and  $\text{C}_6\text{H}_6$ -EtOAc (3:1) (system 5).

PLANT MATERIAL.—The leaves of *P. scutellaria* were collected from Vanuatu by P. Cabalion (ORSTOM). A voucher specimen is deposited in the Herbarium of Botany and Pharmacognosy, Faculté de Pharmacie, Lyon.

EXTRACTION AND ISOLATION OF THE SAPONINS.—Leaves (750 g) were extracted with hot EtOH. The EtOH extract was concentrated, diluted with H<sub>2</sub>O, and extracted with  $\text{CHCl}_3$  to remove lipid material. The H<sub>2</sub>O solution was extracted with EtOAc, then *n*-BuOH. The *n*-BuOH layer was evaporated to dryness to give a crude saponin fraction (21 g). The separation was carried out on a column packed with Si gel 60 (Merck) using a solvent gradient of  $\text{CHCl}_3$ -MeOH-H<sub>2</sub>O (70:2:2 to 70:40:10). Six fractions were collected. Fraction 4 gave polysciasaponin  $P_3$  (30 mg); fraction 6 gave polysciasaponin  $P_1$ . Saponins  $P_1$  and  $P_3$  were further separated on tlc (Si gel) with *n*-BuOH-HOAc-H<sub>2</sub>O (4:1:1) (detection with iodine) to give pure  $P_1$  and  $P_3$ .

METHYLATION BY  $\text{CH}_2\text{N}_2$ .—To an Et<sub>2</sub>O solution of saponin (15 mg), freshly prepared  $\text{CH}_2\text{N}_2$  in Et<sub>2</sub>O was added in excess. The mixture was kept at 37° for 1 h. The reaction product obtained on workup formed a yellow product.

ACID HYDROLYSIS.—The saponin [5 mg in MeOH-H<sub>2</sub>O (2:1) (5 ml)] was refluxed in 0.1 M HCl (5 ml) for 6 h. The aglycone was extracted with  $\text{CHCl}_3$  and identified with an authentic sample by tlc on Si gel and gc-ms after silylation. The aqueous layer was neutralized with *N,N*-octyl-methylamine and evaporated to dryness. The sugars were identified by gc after silylation and by tlc on Si gel using *n*-BuOH-*i*PrOH-H<sub>2</sub>O (5:3:1), or on cellulose using *n*-BuOH-C<sub>5</sub>H<sub>5</sub>N-HCl (5:3:2); compounds were visualized by spraying with aniline hydrogen phthalate and then heating (110°, 10 min).

PARTIAL ACID HYDROLYSIS.—The saponin (5 mg) in MeOH (0.2 ml) and 1 M HCl (0.2 ml) was kept at 70°; after 80 min the mixture was separated by tlc on Si gel with *n*-BuOH-HOAc-H<sub>2</sub>O (4:1:1) to obtain the partially hydrolyzed products.

ENZYMATIC HYDROLYSIS.—The enzymatic hydrolysis was carried out with 1 ml of saponin solution (0.5 ml, pH 6) and 5 mg of  $\beta$ -glucosidase (Extrasynthèse EC 3.2.1.21 300 U/mg) or 0.3 ml of  $\beta$ -glucuronidase (Merck Art. 4114, 12 U/ml, from *Helix pomatia*) during 24 h at 37°. After extraction with  $\text{CHCl}_3$  (sapogenin), glucose or glucuronic acid was identified in the aqueous phase and oleanolic acid in the organic layer.

ALKALINE HYDROLYSIS.—Alkaline hydrolysis was performed at 80° for 3 h with 5 mg of saponin in 1 M NaOH (5 ml). After acidification by 1 M HCl (pH 5), monodesmoside was extracted with *n*-BuOH.

POLYSCIASAPONIN  $P_1$  [**8**].—White powder: mp 263°; tlc  $R_f$  0.22 (system 1), 0.03 (system 2), 0.02 (system 3); fabms  $m/z$  1117 [ $\text{M} - \text{H}$ ]<sup>-</sup>, 955

$[M - H - 162]^-$ ;  $^{13}C$  nmr ( $C_5D_5N$ )  $\delta$  89.5 (C-3), 122.10 (C-12), 144.1 (C-13), 176.3 (C-28), 104.4 (C-1'), 104.7 (C-1''), 104.6 (C-1'''), 95.5 (C-1''').

POLYSCIASAPONIN  $P_3$  [6].—Amorphous powder: mp 260°; tlc  $R_f$  0.38 (system 1), 0.09 (system 2), 0.06 (system 3); fabms  $m/z$  955  $[M - H]^-$ , 793  $[M - H - 162]^-$ . Enzymatic hydrolysis of the methylate derivative of  $P_3$  afforded oleanolic acid, identified by comparison with an authentic sample.

ACID HYDROLYSIS.—Polysciasaponins  $P_1$  and  $P_3$  gave oleanolic acid identified by tlc [ $R_f$  0.71 (system 4), 0.56 (system 5)] and by gc-ms after silylation: ms  $m/z$   $[M]^+$  600,  $[M - COOSi Me_3]^+$  483,  $[M - 280]^+$  320,  $[M - 321]^+$  279,  $[M - 397]^+$  203,  $[M - 411]^+$  189. The sugars were identified by tlc and by gc after silylation. The sugar components were Glc-GlcA (3:1) for saponin  $P_1$ , Glc-GlcA (2:1) for saponin  $P_3$ .

PARTIAL ACID HYDROLYSIS.— $P_1$  gave  $P_2$ ,  $P_3$ ,  $P_5$ , and  $P_7$ ;  $P_3$  yielded  $P_5$  followed by  $P_7$ .

ALKALINE HYDROLYSIS.— $P_1$  afforded  $P_2$ ;  $P_3$  gave  $P_5$ .

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